



Electrical and mechanical characterization of a flexible conducting composite

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Abstract Composite films made of poly(vinylidene fluoride) (PVDF) and conducting particles of carbon black (CB) were prepared using a hot press. Using different volume fractions of CB filler, electrical properties of the samples were analyzed with current–voltage ($I \times V$) measurements and impedance spectroscopy. To help the discussion, percolation theory and simulation circuits based on Colie diagrams were used. The percolation threshold was found at 3 vol% of CB. At these volume fractions of filler, scanning electron microscopy (SEM) images showed connected particles in the polymer matrix, while in the higher volume content of filler, the particles started to form clusters. Furthermore, it was observed that increasing volume fractions up to 3 vol%, the Young's modulus and the tensile strength of the composite film were higher than that of the pure polymer, although the rupture strain was decreased. Composite films with 3 vol% of CB showed optimized electrical and mechanical properties and may be useful as an electrostatic dissipater.

Keywords Conductivity · Mechanical properties · Composite · Carbon black

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Introduction

In general, polymeric materials are flexible and have mechanical resistance, but they have low electrical conductivity. However, for some specific applications, a flexible material with non-negligible electrical conductivity is needed, as in the case of antistatic packaging [1, 2]. To overcome these problems, conducting polymeric composites were fabricated, including semiconductor or conducting particles into the polymer matrix. The resulting material will combine with the mechanical properties and the process ability of the polymer with the electrical property of the conducting filler [1–7].

The composite sample can be obtained by melting the polymer matrix and incorporating the filler or including the filler during the polymerization of the matrix [8]. The properties of the composite depend on the nature of the inclusions and the matrix [9, 10]. An interesting study is the percolation threshold, which indicated the limit of particle/matrix fraction rates where the insulating material became a conductor, i.e. the distribution of the particles in the matrix allowed at least one electrical flux path [6, 11–13].

In the present work, composite films were obtained by hot pressing the mixture of PVDF in powder form with carbon black (CB) particles. The decision to use these materials was made because PVDF is a thermoplastic polymer with good mechanical, optical and thermal properties and is an insulator [14]. On the other hand, CB has been used as conducting filler in polymeric composites, as dye inks, as catalyst in batteries and fuel cells, and as an additive to enhance the mechanical resistance of composites [1, 4, 5, 7]. The main contribution of the present work on flexible conducting composites of ease processing is the correlation between the electrical property of the composite, studied by equivalent circuit, and its mechanical properties to propose a better composition to be used as an electrostatic dissipater.

Experimental

Material

PVDF α -phase in powder form (SOLEF 1008/1001) with 1.78 g/cm³ density was provided by Solvay. CB particles with 50 nm diameter, 1000 m²/g of area and density equal to 1.80 g/cm³ were purchased from Union Carbide.

Composite preparation

CB particles were mixed with PVDF powder in mortar and macerated. Several particles/polymer volume fractions were used, such as 0, 0.5, 1.0, 3.0, 5.0 and 7.0. Each mixture was placed between two polyamide films (Kapton) and pressed at 185 °C for 2 min with 7.0 MPa, using a temperature-controlled hydraulic press (Marconi model MA 098/A). After pressing, the composite film was quickly

immersed in ice water for quenching. Aluminum electrodes were evaporated onto both sides of the composite film for electrical characterization.

Characterization

Conductivity measurements along the surface of the composite film and along its thickness direction were carried out using a current–voltage source (Keithley model 2657A). Field emission electron microscopy (FEG-SEM) was realized using a JEOL model 7500F at Institute of Chemistry, UNESP/Araraquara, and, for the impedance spectroscopy, a Multi Autolab PGSTAT 101 Metrohm with electro-chemical impedance modulus FRA2 was used. The experimental data were adjusted using the EIS Spectrum Analyzer Software. The stress tests were carried out according to ASTM D 1708—10 standard, using a Universal test machine model WDW-300E with 12.5 mm/min speed and 150 kN loading cell.

Results and discussion

Electrical conductivity

The enhancement of DC conductivity of the composite samples with different volume fraction rates of CB was analyzed along the surface and thickness directions. In the surface measurements, there was a considerable increase of the conductivity from 3 vol% of CB, as shown in Fig. 1, indicating that between 1 and



Fig. 1 Electrical conductance DC along the surface of PVDF/CB composites as a function of volumetric fraction of CB particles

3 vol% the material was no longer an insulator and started to be a conducting material. The behavior of the electrical conduction observed in the PVDF/CB composite was in agreement with the results obtained by other authors [15-17].

In this range of CB content, the conductance increased about 5 orders of magnitude, indicating an increasing conductivity of the material with the increase of conducting filler until the percolation threshold. The small increase of the conductivity after the critical concentration for percolation may be attributed to the better contact between the continuous electrical paths, since a small potential barrier allows charge conduction by hopping [7] or tunneling [15], or it may have been due to the enhancement of conducting channels until reaching the saturation when the localized thermal effects can hinder electrical conduction.

The result obtained along the thickness direction was similar, as shown in Fig. 2. The percolation threshold was also at 3 vol%, indicating an isotropic material regarding the CB distribution into the PVDF matrix.

FEG-SEM microscopy

Figure 3a shows CB particles forming small clusters separated by the PVDF matrix. In a better view, in Fig. 3b–d, amplified from Fig. 3a, CB particles can be seen very close to each other through the insulator matrix, starting to form the electrical flux path. The analogous result was observed by Li et al. [18].

The micrograph shown in Fig. 4e, f is related to the 3 vol% of CB into the matrix. The increasing CB content, as expected, made the clusters closer to each other, creating an electrical flux path in the matrix. Consequently, the electrical



Fig. 2 Electrical conductivity DC along the thickness direction of PVDF/CB composites as a function of volumetric fraction of CB particles



Fig. 3 FEG-SEM images of composite with 1 vol% of CB content. The *arrows* show aggregates. The *large arrow (in gray)* indicates the aggregate that has been amplified



Fig. 4 FEG-SEM images of composite with 3 vol% of CB content

conductivity was increased because the electron motion became easier. This means the percolation was reached.

For higher CB content, the particles' aggregate dimensions were more than 50 μ m, and in this case, the mechanical properties of the film could be reduced (Fig. 5).

Impedance spectroscopy

The use of equivalent circuits was quite common in the study of impedance behavior of a sample. This was a simple way to analyze electrical properties of a material, which can be understood as capacitors and resistors associated in different ways to explain the charge conduction and storage behavior [19, 20].

According to simulations, the equivalent circuit, which represented the electrical behavior of a percolated composite, was composed by two resistors and one capacitor associated, as shown in Fig. 6. R_1 was the contact resistance (interface electrode sample) and was used to measure the impedance. R_2 represented the conductive phase (CB), and C_1 , in parallel, represented the capacitive effect of the polymer matrix. There were two distinct behaviors for the electrical charge: it should be storage in the matrix (interface charge), or it would flow through the conducting path as effective electrical current.

For composites with 7 vol% of CB, the adjustment was perfect, as shown in Fig. 7d. However, for 3 vol% (Fig. 7a) and 5 vol% (Fig. 7b), the capacitor needed



Fig. 5 FEG-SEM images of composite with 5 vol% of CB content



Fig. 7 Simulation using equivalent circuits—composites above the percolation threshold. Composites \blacktriangleright with **a** 3 vol%, **b** 5 vol%, **c** 7 vol% (simulated CPE) and **d** 7 vol% (simulated with capacitor) of CB content

Deringer

Fig. 8 Simulation with equivalent circuits—composites below the percolation threshold: a pure PVDF and b PVDF with 1 vol% of CB content

to be changed for a constant phase element (CPE). The CPE has the value of n, near to unit, indicating that it was approximately a charge storage (capacitor) device. It was possible to note that CPE became a perfect capacitor as the CB content increased. In the simulation of the equivalent circuit for a composite with 7 vol%, the value of n was equal to 0.96, and the CPE could be substituted for a capacitor and resistor (Fig. 7d).

For PVDF films or composites with a CB concentration lower than the percolation threshold, the results were similar. The adjustments became harder, and the equivalent circuit was like a capacitor associated, in parallel, with a CPE. In this case there was no conduction, and the charge was essentially storage. As shown in Fig. 8a, b, the adjustment of dielectric loss was very hard. In these cases, the use of Lock-In amplifier and current discharge techniques was better [21].

The electrode diameter, and film thickness, of the samples is shown in the Table 1. These values, together with the adjusted R_2 value, gave the conductivity σ of the composite, using Eq. (1).

$$\sigma = \frac{iA}{Vl} \tag{1}$$

where i is the electrical current intensity, V is the applied voltage, A is the electrode area and l is the sample thickness.

Table 1 Diameter andthickness of the films obtained	Sample	Thickness (µm)	Diameter of electrode (mm)
	PVDF	100	8.5
	PVDF/CB 1 vol%	70	10.0
	PVDF/CB 3 vol%	90	8.5
	PVDF/CB 5 vol%	120	8.5
	PVDF/CB 7 vol%	270	8.5

Fig. 9 Stress-strain tests

The values of σ obtained by Eq. (1) were in agreement with the values obtained by $I \times V$ measurement. For example, the conductivity was around 1×10^{-4} S/cm for the composite with 3 vol%.

Stress-strain tests

Figure 9 shows the results of the stress–strain tests of PVDF/CB composites in the range of 0–7 vol% of CB. In general, the rupture strain of composites decreased as the CB content was increased. For composites with 1 and 3 vol%, the tensile strength was higher than the pure PVDF, but for 5 and 7 vol%, this value was lower when compared with PVDF; indicating that until 3 vol%, the CB was a reinforcing filler since it must restrict the flow of the chains by localized interactions [8, 10, 18].

Table 2 shows the mechanical parameters calculated from Fig. 9. When the CB content was high, the coupling of large clusters of CB restricted its mechanical

Volumetric fraction of CB (%)	Young's modulus (GPa)	Rupture strain (%)	Rupture stress (MPa)	Limit of tensile strength (MPa)	Yield strength (Mpa)
					(
0	1.06	120.0	21.6	28.3	7.2
1	1.21	7.0	40.0	42.0	14.0
3	1.46	3.5	30.0	30.0	13.0
5	1.18	3.0	23.0	23.0	11.0
7	1.40	1.4	16.0	16.0	5.2

 Table 2
 Tensile properties of films

strength. Although the Young's modulus of composite with 7 vol% was higher than the 5 vol%, the other parameters were lower, and the sample was more fragile. The formation of neck during the deformation test was observed only in pure PVDF.

Conclusions

PVDF/CB films with different CB contents were obtained by mixing the powder of both components with subsequent hot pressing. The electrical conductivity of the new material increased as the CB content was increased, allowing the determination of the percolation threshold of the composite samples, which occurred around 3 vol% of filler.

The composite was isotropic, in regard to the electrical property, since the percolation threshold appeared at the same volume fraction of filler in both directions of measurement. Pure PVDF film and composite under percolation threshold behaved as a capacitor, restricting the current flow and working as a charge storage device. Over the percolation, there are two possibilities: there may be an accumulation of charges on the polymer interface, or the charges could flow along the conducting channels (percolated CB phase).

Mechanical properties of the composite could be enhanced using low volume fraction rates of CB. The value of 3 vol% appeared as a better composition, and the composite showed a potential application as an electrostatic dissipater.

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