



Research articles

Towards enhancing the magnetic properties by morphology control of ATiO₃ (A = Mn, Fe, Ni) multiferroic materialsR.A.P. Ribeiro^a, L.H.S. Lacerda^a, E. Longo^b, J. Andrés^{c,*}, S.R. de Lazaro^a^a Department of Chemistry, State University of Ponta Grossa, Av. General Carlos Cavalcanti, 4748, 84030-900 Ponta Grossa, PR, Brazil^b CDMF-UFSCar, Universidade Federal de São Carlos, PO Box 676, 13565-905 São Carlos, SP, Brazil^c Department of Analytical and Physical Chemistry, University Jaume I, Castelló 12071, Spain

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ABSTRACT

Magnetic materials are of great interest due to their widespread applications in various technologies. Precise control of different aspects of the morphology and magnetic properties is a key aspect in establishing their relationship with material shapes and sizes. In this work, the density functional theory (DFT) is applied to investigate the surface structure and magnetic properties of low-index (1 1 0), (1 0 1), (1 0 0), (0 0 1), (1 1 1), (0 1 2) surfaces of ATiO₃ (A = Mn, Fe, Ni) multiferroic materials. In particular, a theoretical approach, based on the Wulff construction and magnetization density (M) index, is applied to infer the relationship between the morphology and surface magnetism. The results indicate that the magnetic properties of ATiO₃ materials can be controlled by the presence of exposed surfaces. In particular, the (0 0 1) and (1 1 1) surfaces are found to be adequate to enhance the superficial magnetism. The underlying reasons behind such mechanism are discussed from the perspective of uncompensated spins and charge-transfer processes along the exposed surfaces. This investigation suggests that the morphology-oriented ATiO₃ materials are good candidates for superior multiferroic applications.

1. Introduction

Theoretical and experimental studies involving magnetic materials combine a broad range of synthetic methods, and characterization techniques applying physics, chemistry, and materials science. The functionality of materials is strongly related to their structure, size, and shape. In this context, the unusual phenomena are associated to the emergence of quantum effects. In particular, the magnetic properties are very sensitive to the morphology due to the dominating role of anisotropy in magnetism when one or more dimensions are reduced [1–4].

The appearance of exposed surface and interface plays a fundamental role in the control of material properties. In particular, this is a key factor that exhibits intriguing magnetic properties at reduced dimensions. The exposed surface presents reduction of metal coordination, creating some vacancies along the exposed surfaces. This reduces the hopping of electrons from site to site, increasing the Coulomb interactions between the electrons. This also enhances the tendency of appearance of superficial magnetism, mainly due to the presence of uncompensated spins originated from the confinement quantum effects [2,5–7].

In the last few years, the combination between the theoretical and experimental techniques has allowed the design of functional materials where the morphological modulation is considered to be a fundamental approach to control several degrees of freedom involved in different physical and chemical processes. Different applications, such as catalysis, gas sensing, energy conversion and storage, and magnetoresistance depend on the surface arrangement at the atomic level [8–11]. The magnetic properties are also affected by the chemical composition, shape, size, and the local details at the core and surface of the sample. These are also dependent on the interactions with the surrounding environment that enables the existence of superparamagnetism, giant magnetic resistance, and superior multiferroic properties [12–16].

Magnetic materials are promising candidates for use in of spintronic technology, where both the electrical and magnetic degrees of freedom are used [2,4]. Numerous experimental investigations have discussed the dependence of the surface morphology and unusual magnetic properties for different materials, such as Co₃O₄, Mn_xO_y, Fe_xO_y, CoFe₂O₄, and MnS [9,17–31].

Our recent efforts are devoted to developing an innovative and alternative method to relate the magnetic properties with crystal morphology using open-shell density functional theory (DFT), and Wulff

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construction. Under this approach, the magnetization density index (M) is evaluated by means of the uncompensated spins along the exposed surfaces, enabling the modulation of crystal morphology to obtain superior magnetic properties as observed for Co_3O_4 and MnTiO_3 [32,33].

In this study, we present a theoretical investigation to control the magnetic properties of ATiO_3 ($A = \text{Mn, Fe, Ni}$) multiferroic materials depending on the morphology, focusing on the description of uncompensated spins, surface exposed undercoordinated cations, and shape-oriented superficial magnetism.

2. Theoretical method and computational procedure

In this work, we calculate the surface energy (E_{surf}) by means of DFT calculations using hybrid PBE0 functional and CRYSTAL14 code [34,35]. The geometry of the (1 0 0), (0 0 1), (1 1 0), (1 0 1), (0 1 2), (1 1 1) surfaces are described using the slab construction with a thickness large enough to ensure full relaxation of the surface atoms and convergence of the surface energy. After completing the corresponding optimization process and thickness convergence test, the repeat units {number of layers in the slab} are selected as follows: for TiO_3A {18 layers} and (0 0 1) surface, $\text{O}_3\text{A}_2\text{Ti}_2\text{O}_3$ {40 layers} for (1 1 0) surface, OTiOAO {30 layers} for (1 0 1) surface, $\text{O}_2\text{A}_2\text{O}_2\text{Ti}_2\text{O}_2$ {30 layers} for (0 1 2) surface, $\text{O}_2\text{A}_2\text{Ti}_2\text{O}_4$ {40 layers} for (1 0 0) surface, and $\text{OTiOAO Ti}_2\text{OAO}$ {50 layers} for (1 1 1) surface. These layers were found to be sufficient for the convergence. In the course of the geometrical optimization, we do not place any constraints on the atoms except for the conservation of the original crystal symmetry in the two dimensions parallel to the surface.

It is important to note that (0 0 1), (1 0 0), (1 0 1), (0 1 2) and (1 1 1) surfaces are polar; therefore, the slab models used are differently terminated, while (1 1 0) is non-polar. Although different surface terminations may occur, only those giving rise to a minimal dipole moment are considered, as presented in Fig. 1.

In our study the surface energy (E_{surf}) is assumed to be the relaxed surface cleavage energy (E_{cleav}). Firstly, we introduce the concept of unrelaxed E_{cleav} as the required energy to cut the bulk along the selected plane.

$$E_{\text{cleav}}^{\text{unrx}} = \frac{(E_{\text{slab}}^{\text{unrx}} - nE_{\text{bulk}})}{2A}$$

Here, $E_{\text{slab}}^{\text{unrx}}$ and E_{bulk} correspond to the total energies for the unrelaxed slab model and the bulk unit, respectively, whereas n and A represent the number of bulk units used in the slab construction and the surface area, respectively. For polar surfaces, a relaxation process of both Z^+ and Z^- terminations is performed; then the $E_{\text{relax}}(Z^+|Z^-)$ is calculated as the difference between the total energies for relaxed and unrelaxed slabs, as follows:

$$E_{\text{relax}}(Z^+|Z^-) = \frac{(E_{\text{slab}}^{\text{unrx}} - E_{\text{slab}}^{\text{relax}})}{2A}$$

being the E_{surf} the mean value for both terminations:

$$E_{\text{surf}} = E_{\text{cleav}}^{\text{relax}}(Z^+|Z^-) = E_{\text{cleav}}^{\text{unrx}} - E_{\text{relax}}$$

In all calculations, A, Ti, and O centers were described from all-electron atomic basis sets composed of Gaussian type functions denominated by 86-411d41G, 86-51(3d)G, and 8-411, respectively [36–38]. It is important to recognize that the electron correlation effects are important in manganese materials. However, we are confident in our results because the computed bulk structural properties and band gaps for ATiO_3 are in agreement with the experimental results [39]. The diagonalization of the Fock matrix was performed at adequate k-point grids (Pack-Monkhorst) in the reciprocal space [40]. The thresholds controlling the accuracy of the Coulomb and Exchange integral calculations were controlled by five thresholds set to 7, 7, 7, 7, and 14; respectively. The irreducible Brillouin zone (IBZ) was represented by a

number of sampling points, which were chosen as $(8 \times 8 \times 8)$ and (4×4) for the bulk and the slab, respectively. As a result, there are 65 k-points in the bulk IBZ, and 4 k-points in the slab IBZ. The convergence criteria for mono- and bi-electronic integrals were set to 10^{-8} Hartree. The RMS gradient, RMS displacement, maximum gradient, and maximum displacement were set to 3×10^{-5} , 1.2×10^{-4} , 4.5×10^{-5} , and 1.8×10^{-4} a.u., respectively.

The classical Wulff construction is employed where the energy of the crystal surface (E_{surf}) determines the equilibrium morphology. A simple relationship between (E_{surf}) and the distance in the normal direction from the center of the crystallite allows to find the ideal morphology. By tuning the surface energies of the different facets, the available morphologies can then be obtained [11,41–49].

The magnetic properties associated with the different morphologies were calculated using the spin density (D_μ) index of a given surface. This is related to the magnetic moment (μ_B) per unit cell area (A). In addition, the combination of the polyhedral representation, $c_{(hkl)}$, derived from the Wulff construction (with D_μ), enables us to predict the total magnetization density (M) index of a given morphology as follows [32]:

$$D_\mu = \frac{\mu_B}{A}$$

$$M = \sum c_{(hkl)} \cdot D_\mu^{(hkl)}$$

To link the obtained morphologies with the ideal shapes, the polyhedron energy (E_{pol}) was calculated by summing the contribution of each surface to morphology ($c_{[hkl]}$) and its E_{surf} , as follows:

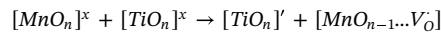
$$E_{\text{pol}} = \sum c_{(hkl)} \cdot E_{\text{surf}}^{(hkl)}$$

where, $c_{(hkl)}$ is the “weight” or percentage contribution of the surface area to the total surface area of the polyhedron, and $E_{\text{surf}}^{(hkl)}$ is the surface energy of the corresponding surface. This methodology provides a simple relationship between the different morphologies, enabling an atomic level understanding of thermodynamic process involved in morphological modulations [48,50].

3. Results and discussion

3.1. MnTiO_3

In Table 1 the calculated E_{surf} values for MnTiO_3 are summarized. The resultant magnetic moment along the exposed surface and D_μ is also shown. In this case, it was observed that the stability order is (1 1 0) > (0 1 2) > (1 0 1) > (1 0 0) > (0 0 1) > (1 1 1), while the calculated magnetic moment along the different surfaces showed an unusual spin population compared to the bulk value. In particular, the polar (0 0 1) and (1 1 1) surfaces showed the highest values of D_μ . This is in agreement with the undercoordinated species exposed in such surfaces. In our previous work, the superficial magnetism was increased from the charge transfer mechanism between the neighboring clusters. This followed the Kroger-Vink notation, where a charge accumulation on superficial $[\text{TiO}_n]^\times$ clusters for (0 0 1) and (1 1 1) surfaces was found using the equation below:



Wulff construction were developed using the calculated E_{surf} values (Table 1) to conduct the morphological modulation of MnTiO_3 with a view to attaining superior magnetic properties. The obtained shapes with highest values of M are shown in Fig. 2. The vacuum ideal morphology proposed by MnTiO_3 exhibited a corner-truncated cylindrical shape that predominantly exposed the (1 1 0) surface, and to a minor extent, the (0 1 2) and (0 0 1) surfaces. Despite the higher value of E_{surf} for (0 0 1) surface, the ideal morphology of MnTiO_3 showed an extent of this surface, suggesting the existence of an intriguing magnetic behavior due to the exposure of unusual magnetic Ti species. Indeed, the

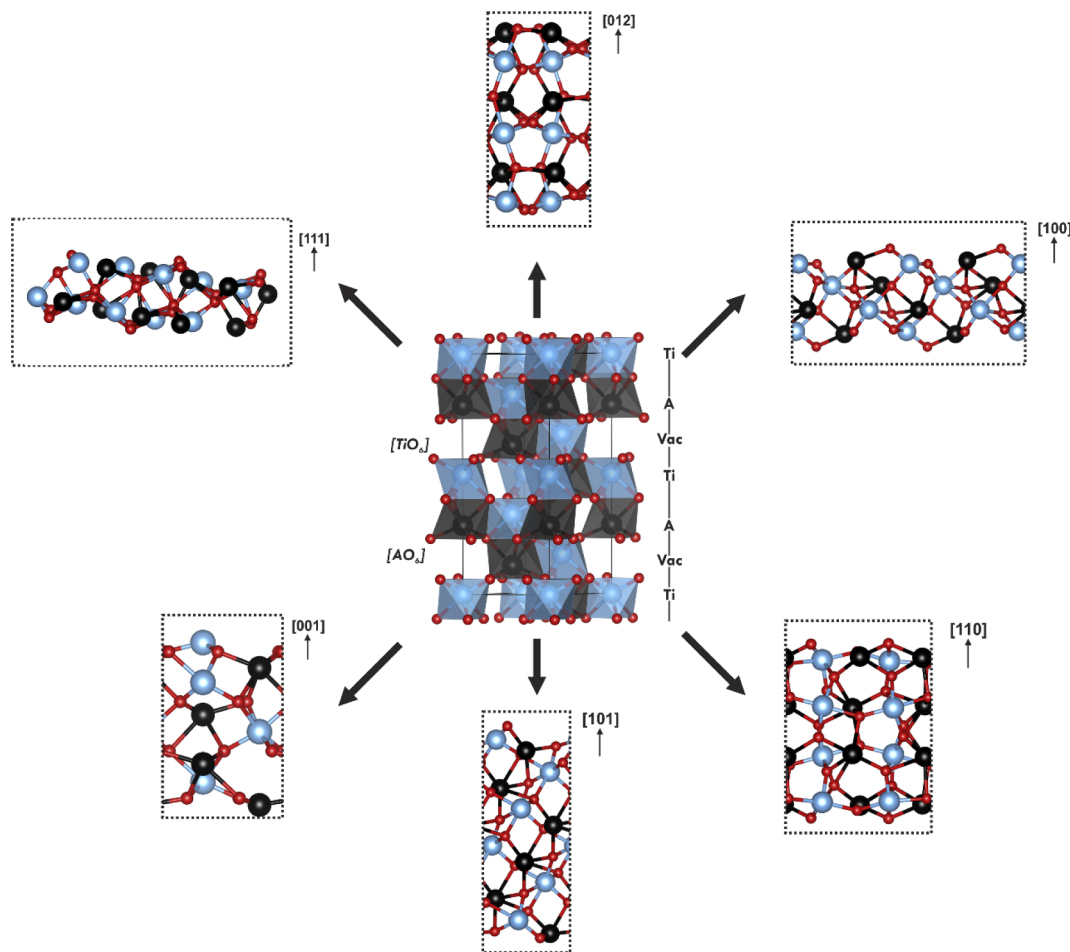


Fig. 1. Crystalline unit cell of $ATiO_3$ ($A = Mn, Fe, Ni$) materials and its surface slabs. The black, blue, and red balls represent A (Mn, Fe, Ni), Ti , and O ions, respectively. The dashed lines represent the lattice vector used in the calculations. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

Table 1

Surface energy (E_{surf}), area (A), magnetic moment and magnetization density ($D\mu$) calculated for the $MnTiO_3$ surfaces.

Surfaces	E_{surf} ($J m^{-2}$)	A (nm^2)	Magnetic Moment (μ_B)			$D\mu$ ($\mu_B nm^{-2}$)
			Z^+	Z^-	Total	
(1 0 0)	2.07	0.710	0	0	0	0
(0 0 1)	2.64	0.235	-0.50	-4.40	-4.90	20.87
(1 1 0)	0.70	0.410	0	0	0	0
(1 0 1)	2.02	0.249	4.70	-4.70	0	0
(0 1 2)	1.98	0.284	-0.20	0	-0.20	0.71
(1 1 1)	3.46	1.252	0.50	0.30	0.80	0.64

increased superficial magnetism for $MnTiO_3$ morphology can be modulated by tuning the E_{surf} values for (0 0 1) and (1 1 1) surfaces.

In the reaction path (A) of Fig. 2 a polyhedral that exposes mainly (1 1 1) surface (with a few extent of both (0 1 2) and (0 0 1) surfaces) was obtained controlling the ratio between E_{surf} values for both (1 1 0) and (1 1 1) surfaces. In this path, a superior M value ($M = 6.51$) was found, which is attributed to the exposure of both (0 0 1) and (1 1 1) surfaces that exhibit an increased magnetic moment (Table 1).

On the other hand, in the path (B) of the Fig. 2, the cylindrical shape (containing only (1 1 0) and (0 0 1) exposed surfaces) was obtained controlling the ratio between the E_{surf} values. This resulted in an increased superficial magnetism ($M = 6.48$) due to the superior contribution of (0 0 1) surface in comparison to the ideal morphology ($M = 5.53$). Recently, Wang *et al*[51] obtained $MnTiO_3$ nanodiscs that

preferentially exposed (0 0 1) surfaces with a few extent of (1 1 0), which can be compared with the theoretical prediction of nanodisc morphology by controlling the E_{surf} ratio for these surfaces, resulting in superior magnetic properties ($M = 12.76$), which are in agreement with the experimental results [51].

In short, our results suggest that in case of multiferroic shape-oriented $MnTiO_3$, superior multiferroic properties can be obtained by controlling the exposure of (0 0 1) and (1 1 1) surfaces due to the presence of reduced $[TiO_n]$ clusters.

3.2. $FeTiO_3$

Here the E_{surf} values of $FeTiO_3$ are calculated, and the spin densities along the surfaces are investigated, as presented in Table 2.

A different stability order was observed in comparison to $MnTiO_3$. Here, the thermodynamic stability order of the surfaces was: (1 1 0) > (0 1 2) > (1 0 0) > (0 0 1) > (1 0 1) > (1 1 1). This shows a change between (1 0 0), (0 0 1) and (1 0 1) surfaces in comparison to $MnTiO_3$. This fact can be attributed to the chemical bond anisotropy along the surface—once the covalent character is distinguished along the Fe-O and Mn-O bonds. In addition, we noted an increase in the E_{surf} value for non-polar (1 1 0) surface in comparison to $MnTiO_3$. This is associated with the increased covalent character of the Fe-O bonds resulting in a higher required energy to create slab model cutting for this surface. In addition, our results suggest a singular spin population for the polar (0 0 1) and (1 1 1) surfaces, resulting from the charge transfer between $[AO_n]$ and $[TiO_n]$ ($n = 2-5$) clusters in order to generate

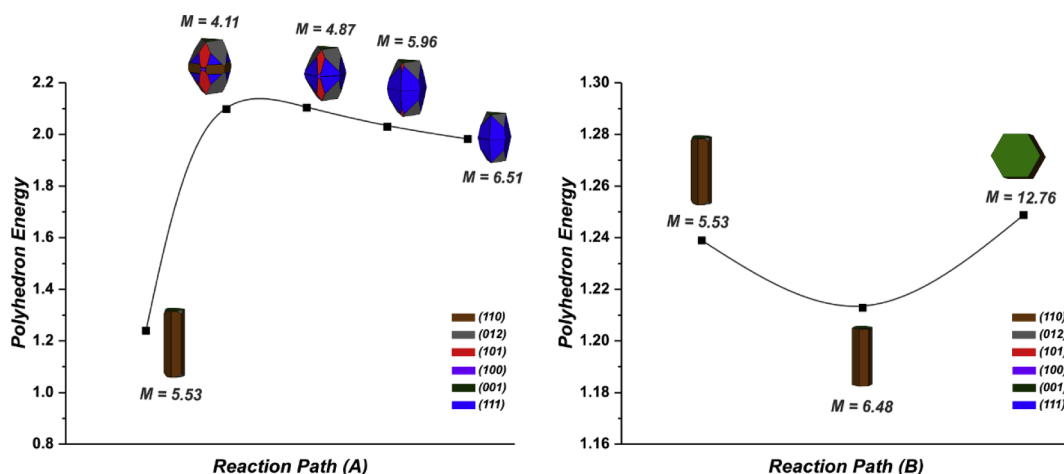


Fig. 2. Schematic representation of the energy profile to obtain different MnTiO_3 morphologies. The intermediate morphologies are obtained by decreasing/increasing the E_{surf} values of the corresponding surfaces involved in the process.

Table 2

Surface energy (E_{surf}), area (A), magnetic moment and magnetization density (D_{μ}) calculated for the FeTiO_3 surfaces.

Surfaces	$E_{\text{surf}}(\text{J m}^{-2})$	$A (\text{nm}^2)$	Magnetic Moment (μ_{b})			$D_{\mu} (\mu_{\text{b}} \text{nm}^{-2})$
			Z^+	Z^-	Total	
(1 0 0)	2.00	0.702	0	0	0	0
(0 0 1)	2.23	0.230	-0.74	-3.71	-4.45	19.33
(1 1 0)	1.57	0.405	0	0	0	0
(1 0 1)	2.25	0.246	3.89	-3.89	0	0
(0 1 2)	1.85	0.280	0	0	0	0
(1 1 1)	3.89	1.237	-0.73	-0.32	-1.04	0.84

reduced $[\text{TiO}_n]$ clusters, as discussed previously.

As for MnTiO_3 , the morphological map of FeTiO_3 was investigated considering the E_{surf} values and the spin density along the surface, aiming to select and predict the morphologies with superior magnetic and multiferroic properties, as presented in Fig. 3.

The vacuum ideal morphology of FeTiO_3 was mainly composed by (1 1 0) and (0 1 2) surfaces with a minor extent of (0 0 1) and (1 0 1), corresponding to a corner and edge-truncated cylindrical shape. The exposure of (0 0 1) surface induces a $M = 4.99$, suggesting that the control of superficial magnetism for FeTiO_3 can be made by controlling the ratio between E_{surf} values for (0 0 1) and (1 1 1), due to the increased spin density along the surfaces. Indeed, in the path (A) of Fig. 3, the control of E_{surf} for (1 1 1) generates different shapes with increased contribution of this surface, resulting in increased superficial magnetism even compared with MnTiO_3 . In this path, the thermodynamic

pathway to obtain superior magnetic properties requires an activation energy associated with the increase of E_{surf} values for (1 1 0) and (0 1 2) surfaces summed to the stabilization of (1 1 1) surface.

On the other hand, in the reaction path (B) of Fig. 3 only the E_{surf} value for (0 0 1) was changed, generating distinct truncated cylindrical shapes with an increased contribution of (0 0 1) surface that exhibits singular magnetic properties along the surface. This results in an increased superficial magnetism moving from the ideal ($M = 4.99$) to the hexagonal cylindrical shape ($M = 12.36$). In comparison to the experimental results for hexagonal nanodisc of MnTiO_3 [51], the predicted shape for FeTiO_3 shows a higher value of unpaired electrons along the exposed surfaces, being an excellent candidate for multiferroic applications.

3.3. NiTiO_3

The calculated energy and magnetic properties of NiTiO_3 are presented in Table 3. For this material, the stability order of the surfaces, based on the E_{surf} values, was: (1 1 0) > (0 1 2) > (1 0 0) > (0 0 1) > (1 0 1) > (1 1 1). This shows the same order observed for FeTiO_3 . This result can also be supported by the increase of the covalent character for both A-O and Ti-O moving from Mn to Ni in ATiO_3 , as it was demonstrated by a QTAIM analysis carried out for bulk ATiO_3 ($A = \text{Mn, Fe, Ni}$) [39]. The two-dimensional slabs, formed by cutting the bulk along the selected surfaces, provoke a chemical anisotropy along the crystalline structure. Therefore, this increase the covalent character of both A-O bonds, moving from Mn to Ni, decreasing the electron mobility.

By analyzing the spin density distribution, it was observed that (0 0 1) and (1 1 1) surfaces show a singular spin density along the

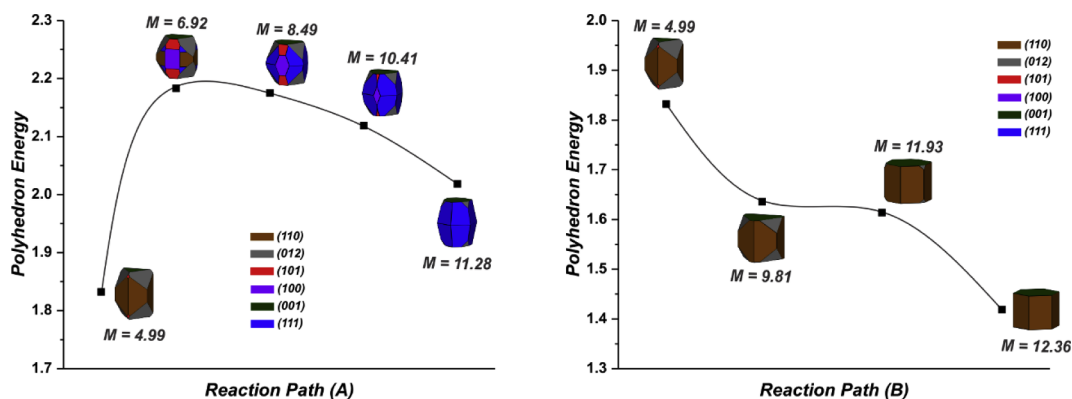


Fig. 3. Schematic representation of the energy profile to obtain different FeTiO_3 morphologies.

Table 3
Surface energy (E_{surf}), area (A), magnetic moment and magnetization density (D_M) calculated for the NiTiO₃ surfaces.

Surfaces	E_{surf} (J m ⁻²)	A (nm ²)	Magnetic Moment (μ_B)			D_M (μ_B nm ⁻²)
			Z ⁺	Z ⁻	Total	
(1 0 0)	2.43	0.691	0	0	0	0
(0 0 1)	2.46	0.220	0	-1.69	-1.69	7.70
(1 1 0)	1.92	0.399	0	0	0	0
(1 0 1)	2.78	0.242	1.68	-1.68	0	0
(0 1 2)	2.38	0.273	0	0	0	0
(1 1 1)	5.64	1.217	0.30	-0.59	-0.29	0.24

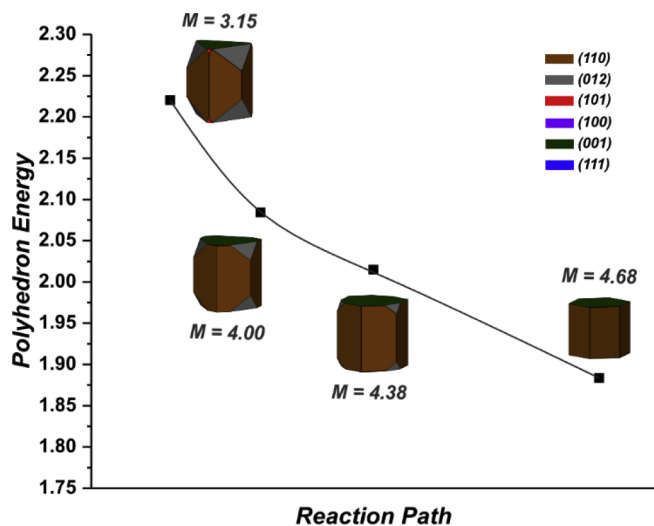


Fig. 4. Schematic representation of the energy profile to obtain different NiTiO₃ morphologies.

exposed surfaces related to the uncompensated spins. However, because of the increased covalent character associated with the Ni-O bonds, the creation of oxygen vacancies along the surface does not induce the charge transfer between [NiO_n] and [TiO_n] clusters ($n = 2-5$). This suggests a higher spin density redistribution in empty 3d orbitals of Ni²⁺, reducing the magnetic moment along the surfaces. Due to the exposure of uncompensated Ni²⁺ spins, (0 0 1) is the best surface to induce superficial magnetism in NiTiO₃.

Considering the E_{surf} values of Table 3 and the Wulff construction, a corner and edge-truncated cylindrical ideal morphology is obtained (Fig. 4). As observed, (1 1 0) and (0 1 2) surfaces are the most stable surfaces, followed by (0 0 1) surface, and a minor extent of (1 0 1) surface. The superficial magnetism calculated for the ideal morphology shows the smallest value ($M = 3.15$) along the ATiO₃ ($A = Mn, Fe, Ni$) series, which is attributed to the lowest spin density along the exposed surfaces. To increase the superficial magnetism in NiTiO₃, the control of (0 0 1), (1 1 0) and (0 1 2) surfaces' E_{surf} values can be adjusted to induce a higher M for the hexagonal nanodisc ($M = 4.68$). This shows that such surface plays a fundamental role in the control of magnetic properties of NiTiO₃. This result confirms the experimental observations associated with the growth mechanism of NiTiO₃ thin-films that expose (0 0 1) surfaces with increased weak ferromagnetism and superior multiferroic properties [52–54].

In this work, we have presented a systematic investigation about the morphological modulations of multiferroic ATiO₃ ($A = Mn, Fe, Ni$) using an alternative and innovative method that suggests that enhanced magnetic properties can be obtained by controlling the exposure of (0 0 1) and (1 1 1) surfaces for MnTiO₃ and FeTiO₃, respectively. On the contrary, for NiTiO₃ the (0 0 1) surface is mandatory.

These results combine an atomic-level description of the uncompensated spins along the exposed surfaces, charge transfer mechanism between the exposed clusters centered in under coordinated cations, as well as thermodynamic manipulation of E_{surf} values in order to select the morphologies with increased magnetization density index. The main novelty of our work is in suggesting the best morphologies for multiferroic applications. It was found that ATiO₃ ($A = Mn, Fe, Ni$) is an excellent candidate for this type of niche applications

Finally, it is important to note that the calculated morphologies are associated to the stabilization and/or destabilization of the exposed surfaces, and the magnetic properties are linked to the presence of excited electronic states. The combination of the structural and electronic disorder at the exposed surfaces introduces different kinds of dangling bonds that are capable of creating under coordinated cations along the exposed surfaces while cutting the bulk of ATiO₃ ($A = Mn, Fe, Ni$). Therefore, the excited electronic states concentrate near the fundamental electronic state with intriguing magnetic properties. This possibility and the mechanism behind the stabilization and characterization of the excited electronic states in this class of materials will be reported in forthcoming studies.

4. Conclusions

In this study, DFT calculations were performed to obtain the surface energies and the corresponding morphologies based on the Wulff construction and spin density. By considering the uncompensated spins of a given surface as well as the pathways between vacuum ideal morphology and a given shape, the magnetic properties of ATiO₃ ($A = Mn, Fe, Ni$) materials are analyzed. By controlling the ratio of the surface energy value, different kinds of morphologies were predicted. In case of MnTiO₃ and FeTiO₃, the results indicate that the control of (0 0 1) and (1 1 1) exposed surfaces lead to shape-oriented materials with increased superficial magnetism associated with the exposure of reduced [TiO_n] species and uncompensated spins. On the other hand, for NiTiO₃, the control of (0 0 1) exposed surface is the best way to select the morphologies with highest superficial magnetism due to the chemical character associated with the Ni-O bonds. The ability to produce ATiO₃ ($A = Mn, Fe, Ni$) nanoparticles with high magnetization is highly desirable because of the remarkable multiferroic applications associated with such compounds. Based on these experimental results reported in the literature to date, we can conclude that by controlling the crystallinity and the exposed surface of the materials, it is possible to influence the magnetization properties of ATiO₃ ($A = Mn, Fe, Ni$) materials. Further investigations aiming at expanding the application of this approach to other materials with potential magnetic properties are currently in-progress in our laboratory.

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